Al-Mg-Mn-Zn (Aluminum-Magnesium-Manganese-Zinc)

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Recently, [2006Ohn1] presented a thermodynamic assessment of this quaternary system, incorporating new results from a few key experiments. Five vertical sections for Mg-rich alloys were computed and compared with the available experimental data.

Ternary Subsystems

Recently, [2007Rag] reviewed the Al-Mg-Mn and Al-Mg-Zn ternary systems. Thermodynamic assessments have appeared recently of Al-Mg-Mn [2005Ohn], Al-Mg-Zn [1998Lia, 2006Ohn2], and Mg-Mn-Zn [2006Ohn3]. The experimental data on the Al-Mn-Zn system were compiled by [1995Vil].

Quaternary Phase Equilibria

Starting with two commercial alloys AZ91HP and AZ61, [2006Ohn1] prepared two alloys with the following composition in mass %: 6.61Al-1.32Zn-0.24Mn and 8.67Al-0.63Zn-0.21Mn. Differential thermal analysis (DTA) and differential scanning calorimetry techniques were employed to identify the thermal arrests. Heating/cooling rates of 1– 5 °C/min were used by [2006Ohn1]. Microstructural and compositional analyses were done with the scanning electron microscope and the energy dispersive x-ray analyzer. [2006Ohn1] emphasized the need for care in interpreting the thermal signals in terms of the phase changes in the alloy.



Fig. 1 Al-Mg-Mn-Zn computed vertical section at 9.5Al-0.84Zn (mass%) [2006Ohn1]

The thermodynamic descriptions of Al-Mg-Mn by [2005Ohn], of Al-Mg-Zn by [1998Lia], and of Mg-Mn-Zn by [2006Ohn3] were adopted by [2006Ohn1], with some improvements based on new experimental results. In the absence of a thermodynamic assessment of the Al-Mn-Zn system, it was described as an extrapolation of the binary parameters.

For the quaternary system, [2006Ohn1] computed five vertical sections and compared them with the available thermal data. These sections are shown in Fig. 1 to 5. In Fig. 1 and 2, the computed liquidus agrees well with the liquid-solid equilibrium compositions determined by



Fig. 2 Al-Mg-Mn-Zn computed vertical section at 9.1Al-1.53Zn (mass%) [2006Ohn1]



Fig. 3 Al-Mg-Mn-Zn computed vertical section at 6.68Al-1.82Zn (mass%) [2006Ohn1]



Fig. 4 Al-Mg-Mn-Zn computed vertical section at 9Al-1Zn (mass%) [2006Ohn1]

[1992Tho]. In Fig. 3 to 5, two thermal signals were observed in all thermal analyses. The signal at the higher temperature does not correspond to the liquidus temperature, but to the phase boundary $(L + Mn_5Al_8)/[L + Mn_5Al_8 + (Mg)]$. Also, the lower signal is below the solidus temperature and corresponds to the precipitation of Mg₁₇Al₁₂(γ) under Scheil (nonequilibrium) conditions of cooling. [2006Ohn1] computed constant temperature contours on the solidus surface near the Mg corner. These were found to show fair agreement with the available experimental data.

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Fig. 5 Al-Mg-Mn-Zn computed vertical section at 14.6Al-0.4Zn (mass%) [2006Ohn1]

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